

Abstract No. over648

**The Electron Density Distribution in a Tetra-nuclear Iron(III) bis( $\mu^3$ -oxo) Butterfly-like Compound,  $[\text{Fe}_4^{\text{III}}\text{O}_2(\text{C}_5\text{H}_9\text{O}_2)_8(\text{C}_6\text{H}_7\text{N})_2]\cdot 2\text{CH}_3\text{CN}$**

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Beamline(s): X3A1

**Introduction:** In our investigation of tri-nuclear oxo-centered iron compounds [1], we encountered an interesting side-product,  $[\text{Fe}_4^{\text{III}}\text{O}_2(\text{C}_5\text{H}_9\text{O}_2)_8(\text{C}_6\text{H}_7\text{N})_2]\cdot 2\text{CH}_3\text{CN}$ . This compound exhibits two equivalent  $\text{Fe}_3(\mu^3\text{-O})$  environments, which are suitable for comparison with the previously studied bonding environments in the trinuclear carboxylates [2].

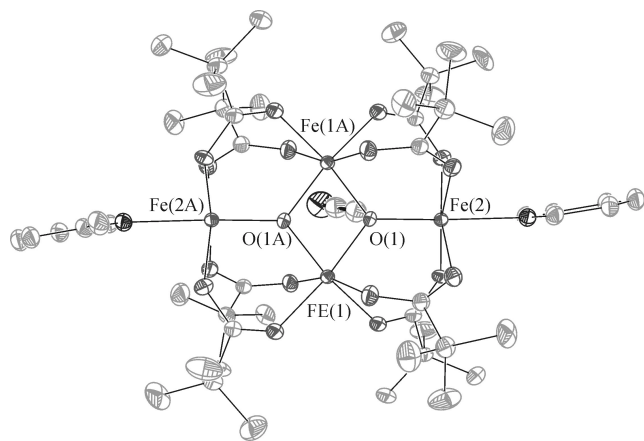
**Methods and Materials:** Synchrotron X-ray diffraction data for the title compound were collected with a Bruker SMART 6000 CCD detector at beamline X3A1 using a liquid He-cooling device to reach an experimental temperature of 16(5) K. 55130 reflections were integrated, giving 20960 unique reflections with an internal agreement of  $R_{\text{int}}=0.025$ . The electron density distribution (EDD) in the complex was modelled employing a multipolar model to give  $R(F^2)=0.029$  for 715 parameters.

**Results:** In contrast to observations in the trinuclear complexes, the EDD in the title complex shows two very significantly different  $\text{Fe}^{\text{III}}\text{-(}\mu^3\text{-O)}$  interactions, the shorter  $\text{Fe(2)-O(1)}$  being very directional with a large accumulation of electron density in the bond, while the longer  $\text{Fe(1)-O(1)}$  interaction is more diffuse (Figure 2). Furthermore,  $\text{Fe(1)}$  is bonded to both  $\mu^3\text{-O}$  atoms, which exhibit an unusually short separation of 2.532(1) Å, see Figure 1. Topological analysis of the experimental EDD reveals a polarization of the valence electrons on the  $\mu^3\text{-O}$  into the  $\text{O(1)-O(1)}$  direction. A similar observation of enhanced interaction between two metal-bridging oxygens has been made previously from low-resolution synchrotron X-ray data [3].

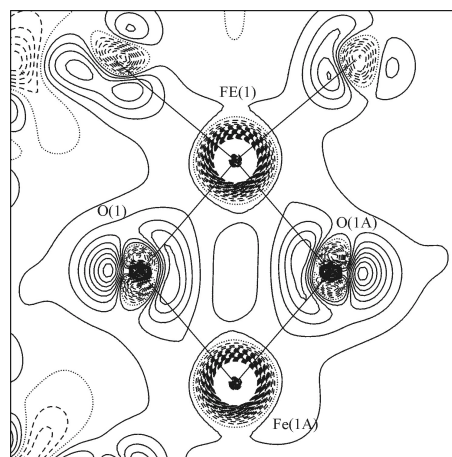
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**References:**

- [1] C. Wilson, *et al.*, *J. Am. Chem. Soc.* (2000), 122, 11370-11379
- [2] J. Overgaard, Ph. D. thesis, Department of Chemistry, University of Aarhus, Denmark, 2001.
- [3] A. F. Jensen, *et al.*, *Inorg. Chem.* (1995), 34, 4244-4252



**Figure 1.** Thermal ellipsoids plots showing 90% probability surfaces.



**Figure 2.** Deformation density in the central  $\text{Fe(1)-O(1)-Fe(1A)-O(1A)}$  region. Contours at  $\pm 0.1 \text{ e}\text{\AA}^{-3}$ .